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Adding the number of protons (12) and the number of neutrons (11) together, reveals the mass of the atom to be 23. Many excited state configurations are possible. A few examples include: 2-7-4, 2-5-6, 1-8-4. They contain different numbers of

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neutrons. $(19.99 \times .909) + (20.99 \times 0.03) +$
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 $1.07) 100$

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ANSWERS To TOPIC 12 Pret is an



CH 41. Butanol or n-butanol 42. [1] Allow
1 credit. Examples of a 1-credit response:

12. 3 43. 36 Review Questions 10. 13. 15.
16. 19. 22. 25. 28. 31. 34. 40. 43. 46. 49.

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52. 55. 1 2 11. 3 14. 3 —c— c- c— H c
c— 18. 21. 24. 30 33. 36. 39. 42. 45. 48.
51. 54. 57. (a) _ Ole + 21gBi (b) _qe +
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Monitoring pH changes during acid-base
titrations (Required Practical 9)

Topic 12 - Acids, Bases and Buffers - A-

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Level Chemistry

A solution was made by dissolving 12.5 g of Na_3PO_4 in 125.00 mL of water. The volume of the resulting solution was 128 mL. Calculate the molarity and molality of the solution.

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$(12.00 \times 9.93) + (13.00 \times 1.07) 100$ Both isotopes contain 6 protons but different numbers of neutrons. Or: An isotope with a mass of 12.00 contains 6 protons and 6 neutrons, while the isotope with a mass of 13.00 contains 6 protons and 7 neutrons.

CHEMISTRY ANSWER KEY 27 Na

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Mg Al $1s^2 2s^2 2p^6 3s^2 3p^1$ Si
 $1s^2 2s^2 2p^6 3s^2 3p^2$ P S $1s^2 2s^2 2p^6 3s^2 3p^4$

Quia

12.02 They contain different numbers of neutrons. $(19.99 \times .909) + (20.99 \times 0.03) + (21.99 \times 0.88)$ 20 $(12.00 \times 9.93) + (13.00 \times 1.07)$ 100 Both isotopes contain 6 protons

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but different numbers of neutrons. Or: An isotope with a mass of 12.00 contains 6 protons and 6 neutrons, while the isotope with a mass of 13.00 contains 6 Na Mg
 $1s^2 2s^2 2p^6 3s^2$

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12 + H₂O → H₂O + + 1- The vertical

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columns of the table should have elements with similar chemical properties. If the elements were arranged by atomic mass, several pairs of elements, such as Ar and K, would be out of position in regards to properties. Both groups follow the octet rule. Group I members achieve an octet by losing an electron,

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Review Book

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Chemistry: The Physical Setting Answer
Key ANSWERS To Topic 1 "MW
Questions 1. 1 2 3 - 3. 2 " " 5- ' 3 o. 2

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7 ' 1 3- 3 9.

Chem Key - Chemistry The Physical
Setting Answer Key ...

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Questions These are the topics a student
should know fairly well in order to skip
Chemistry 122 and enter directly into

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Chemistry 123 or 128. The more difficult topics are marked with an asterisk.

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Contrapositive: If two angles are not
vertical, then they are not congruent.

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